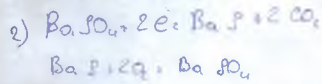
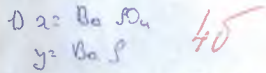
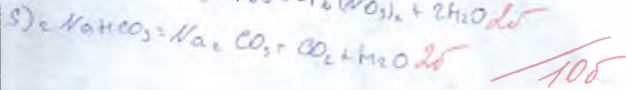
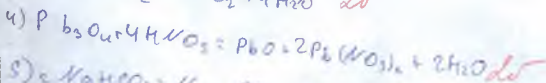
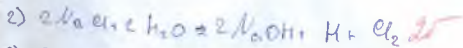
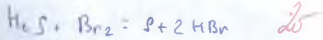


n1

1) Ba



n2



n3



2) $n(\text{Ca}) = n(\text{H}_2) = \frac{0,56}{22,4} = 0,025 \text{ моль}$ 10

$n(\text{Ca}) = \frac{4}{40} = 0,1 \text{ моль}$

$0,1 - 0,025 = 0,075 \text{ моль}$

$m(\text{Ca}) = 0,075 \cdot 40 = 3,2$

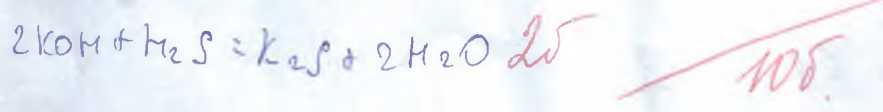
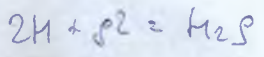
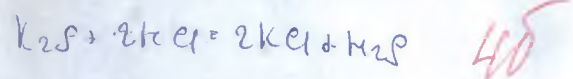
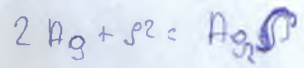
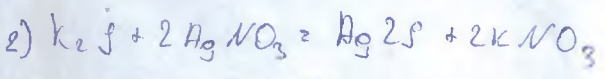
3) $\text{Ca} = 19,2\%$

$\text{CaO} = 80,8\%$

30
 30 / 100

0 1s² 2s² 2p⁶ 3s² 3p⁶ - 21. короче период K₂S 15

15 14



100

v

ответ: 375.